KERALA AGRICULTURAL UNIVERSITY B.Tech. (Food Engineering) - 2011 Admission 1st Semester Final examination – February – March 2012

Cat. No: Base 1104 Title: Engineering Chemistry	Marks: 80 Time : 3 hours
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	Time: 3 hrs.
I. State True / False or Fill up the blanks	(10x1=10)
Precipitation is the primary source of water.	
2. The coal of highest rank is anthracite coal.	
3. Bakelite is a thermoplastic polymer.	
4. The reduction potential of an electrode can be in	creased by decreasing the
concentration of the metal ions in contact with m	netal electrode.
5. R _f = The distance moved by the pure substance The distance moved by the solvent front	
6. Wood → Peat → Lignite → Bituminous coal →	Anthracite coal are the
Progressive transformation of wood to	
7 Corrosion occurs when two dissimilar	metals are electrically connected
and exposed to an electrolyte, the metal higher i	n the Electrochemical series
undergoes corrosion.	
3. The process of reducing frictional resistance between	ween moving /sliding surfaces,
by the introduction of lubricant in between them,	is called
Addition of Sodium hexa meta phosphate to prev	ent sludge and scale formation in
boiler water is called	

10. The phenomenon in which a metal or an alloy exhibit a much higher corrosion resistance than expected from its position in the galvanic series is called

II. Answer any Ten of the following

(10x3 = 30)

- 1. What are the advantages of Zeolite/ permutit process?
- 2. Distinguish between gross and net calorific value.
- 3. Distinguish between thermo plastic resin and thermosetting resin.
- 4. Why does natural rubber need compounding?
- 5. Give a brief account of the working of a concentration cell.
- 6. What are the important corrosion inhibitors? Give examples
- 7. Write a short note on partition chromatography.
- 8. Give a brief account of the classification of lubricants with examples.
- 9. Distinguish between (i) cloud point & pour point (ii) flash point & fire point.
- 10. Define reverse osmosis. What is the principle of reverse osmosis?
- 11. What is the principle of High Pressure Liquid Chromatography (HPLC)?

 What are the important uses?
- 12. Write a short note on Biogas production.

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III. Answer any Six of the following

(6x5=30)

- 1. Give a brief account of the various methods used for the disinfection of water.
- 2. Explain the flue gas analysis by using Orsat apparatus. What is its significance?
- 3. Describe the synthesis of gasoline by Fischer-Tropsch method and Bergius process.

- 4. Differentiate between addition polymerization and condensation polymerization with suitable examples.
- 5. What are the drawbacks of raw rubber and what are the advantages of Vulcanization?
- 6. Distingush between a reversible cell and an irreversible cell.
- 7. Explain the mechanism of wet corrosion.
- 8. Derive Beer- Lambert's law.

IV. Answer any One of the following.

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(10x1 = 10)

- 1. Derive Nernst's equation. What are the applications?
- 2. Explain the methods of controlling corrosion.