KERALA AGRICULTURAL UNIVERSITY

B.Tech.Food Engg. 2013 Admission One Time Re-examination-January 2017

Cat. No: Meen.1203 Marks: 50
Title: Engineering Thermodynamics(2+1) Time: 2 hours

I. Fill up the blanks/True of False

 $(10 \times 1=10)$

- In an irreversible process, there is no loss of heat.(T/F)
- 2. With increase in pressure the enthalpy of dry saturated steam decreases.(T/F)
- 3. A system comprising a single phase is called as -----
- 4. The temperature of air as registered by an ordinary thermometer is called as -----
- The difference between the dry bulb and dew point temperatures is called as -----
- 6. The difference between the superheated temperature and the saturation temperature at the given pressure is called -----
- 7. Gas constant is equal to the -----of two specific heats.
- 8. The latent heat of vapourisation at critical point is equal to -----
- 9. One joule is equal to ----- Nm.
- 10. The temperature at which the volume of a gas becomes zero is called as -----

II. Write short notes/answers on ANY FIVE:

(5x 2=10)

- 1. What is triple point?
- 2. Find the specific volume, enthalpy and internal energy of wet steam at 18 bar. The dryness fraction is 0.9.
- 3. What do you mean by Clausius inequality?
- Define relative efficiency and air-standard efficiency.
- 5. What advantages are obtained if superheated steam is used in steam prime movers?
- 6. State Boyle's law and Charle's law.
- 7. Define equation of state.

III Write answers on ANY FIVE:

 $(5 \times 4 = 20)$

- A carnot engine working between 400°C and 40°C produces 130 kJ of work. Determine

 (a) Engine thermal efficiency (b) Heat added and (c) Entropy changes in the heat rejection process.
- One kg of gaseous CO₂ contained in a closed system undergoes a reversible process of constant pressure. During this process 42 kJ of internal energy is decreased. Determine the work done during this process.
- 3. The volume of a high altitude chamber is 40m³. It is put into operation by reducing pressure from 1 bar to 0.4 bar and temperature from 25°C to 5°C. How many kg of air must be removed from the chamber during the process.
- 4. Discuss briefly about Diesel cycle.
- 5. Discuss in detail about the properties of steam.

- What amount of heat would be required to produce 4.4 kg of steam at a pressure of 6 bar and temperature of 250°C from water at 30°C? Take specific heat of super heated steam as 2.2 kJ/kg K.
- 7. Write short notes on second law of thermodynamics.

IV. Write essay on any ONE

(1 x 10=10)

- Discuss at length about the PVT diagram on phase change describing its significance and its parameters.
- 2. Discuss in detail about Constant volume or Otto cycle with the help of suitable sketches.
